

DEVICE FOR MEASURING IONIC ACTIVITIES

FIELD OF THE INVENTION

5 This invention relates to a device for measuring ionic activities of various ionic components contained in a sample solution, which is favorably employable for simultaneous analysis of ionic components in liquid samples such as a whole blood sample, a serum sample, and a urine sample.

BACKGROUND OF THE INVENTION

15 An ionic activity-measuring device utilizing an ion selective electrode is widely employed for analyzing ionic components in a liquid sample such as a whole blood sample or a serum sample.

20 United States Patent No. 4,571,293 (which corresponds to EP 0 160 997 B1) discloses an ionic activity-measuring device using an ion selective electrode which is illustrated in Fig. 1 of the drawings attached to this specification.

25 In Fig. 1, the ionic activity-measuring device comprises a non-electroconductive support 11, a pair of electrodes each of which comprises a silver metal layer 12a, 12b and a silver halide layer 13a, 13b, a common electrolytic material layer 14, a common ion selective membrane 15, and a common non-electroconductive cover sheet 16 having a pair of openings 17a, 17b for receiving and keeping a sample solution and a reference solution, respectively, each opening being placed over each electrode unit, and having thereon an a bridge member 18 for electrically bridging the sample solution received in one opening 17a and the reference solution received in another opening 17b. Each silver metal layer 12a, 12b has an

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exposed surface 10a, 10b, respectively. By placing probes of a potentiometer 19 on these exposed silver metal surfaces 10a, 10b, the produced electric potential difference can be measured.

5 The ion selective electrode can measure an ionic activity of an ionic component such as H^+ , Li^+ , Na^+ , K^+ , Mg^{2+} , Ca^{2+} , Cl^- , HCO_3^- , or CO_3^{2-} , by employing an appropriate ion selective membrane.

10 United States Patent No. 4,789,435 describes an ion selective electrode assembly comprising plural ion selective electrodes for analyzing plural ionic components such as Na^+ , K^+ , and Cl^- , simultaneously. In the assembly, one of plural ion selective electrodes has an ion selective membrane differing from that of other ion selective electrode in chemical composition.

SUMMARY OF THE INVENTION

20 It is an object of the present invention to provide a device for measuring simultaneously ionic activities of plural ionic components contained in an aqueous sample.

The invention resides in a device for measuring ionic activity which comprises:

25 a block of insulating material having a hollow space therein, a solution-receiving surface area in which a pair of openings for receiving a sample solution and a reference solution separately are provided, said openings connecting with the hollow space, a plurality of solution-supplying surface areas in each of which a pair of
30 openings for supplying outside the sample solution and the reference solution separately are provided, said openings connecting with the hollow space;

35 a bridge member provided on the solution-receiving surface area for electrically bridging the sample solution received in one opening and the reference solution received in another opening;

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a guide member placed in the hollow space which as-
sists to transmit separately the sample solution received
in the opening in the solution-receiving surface area to
the openings on the solution-supplying surface area for
the supplying outside the sample solution and the refer-
ence solution received in the opening in the solution-
receiving surface area to the openings on the solution-
supplying surface area for the supplying outside the
reference solution;

and

a plurality of ion selective electrodes having an
ion selective membrane thereon each of which is placed on
the solution-supplying surface area under such condition
that the ion selective membrane is brought into contact
with the sample solution and the reference solution sepa-
rately.

Preferred devices according to the invention are as
follows.

(1) The device wherein the guide member is a parti-
tion which is placed in the hollow space to guide sepa-
rately the sample solution and the reference solution.

(2) The device wherein the guide member comprises a
pair of porous liquid-transmitting materials placed in
the hollow space, one of which transmits the sample solu-
tion and another of which transmits the reference solu-
tion.

(3) The device wherein the block is in the form of
a rectangular parallelepiped having an upper surface on
which the solution-receiving surface area is arranged,
and other surfaces including a bottom surface and side
surfaces on at least two of which the solution-supplying
surface area is arranged.

(4) The device wherein the block is in the form of
a horizontally extended rectangular parallelepiped having
an upper surface on which the solution-receiving surface
area is arranged, a bottom surface, and side surfaces,

the plurality of the solution-supplying surface areas are arranged on at least one of these surfaces.

(5) The device wherein the block is in the form of a vertically extended rectangular parallelepiped having a upper surface on which the solution-receiving surface area is arranged, and other surfaces including a bottom surface and side surfaces on at least two of which the solution-supplying surface area is arranged.

BRIEF DESCRIPTION OF DRAWINGS

Fig. 1 is a schematic view of a representative structure of a conventional ionic activity-measuring device.

Fig. 2 is a schematic view of a representative structure of the ionic activity-measuring device according to the invention.

Fig. 3 is a schematic view of the device of Fig. 2 in a disassembled form.

Fig. 4 illustrates an inner structure of the block of the device of Fig. 2.

Fig. 5 is a horizontal section of an inner structure of a block of different structure.

Fig. 6 is a schematic view of an inner structure of a block of different structure.

Fig. 7 is a partly enlarged view of a vertical section of a solution-supplying opening.

Fig. 8 is a schematic view of an ionic activity-measuring device of the invention having a different structure.

Fig. 9 is a schematic view of an ionic activity-measuring device of the invention having a different structure.

Fig. 10 is a schematic view of an ionic activity-measuring device of the invention having a different structure.

Fig. 11 is a schematic view of an ionic activity-measuring device of the invention having a different structure.

Fig. 12 is a schematic view of an ionic activity-measuring device of the invention having a different structure.

Fig. 13 is a schematic view of a frame for encasing the ionic activity-measuring device of the invention.

Fig. 14 is a schematic view illustrating manufacture of the block of the device of Fig. 10.

DETAILED DESCRIPTION OF THE INVENTION

The ionic activity-measuring device of the present invention is further described by referring to the fig-

Fig. 2 is a schematic view of a typical structure of the ionic activity-measuring device according to the invention, and Fig. 3 is a schematic view of the device of Fig. 2 in a disassembled form.

In Fig. 2 and Fig. 3, an ionic activity-measuring device 20 is composed of a block 21 of rectangular parallelepiped which has a hollow space in its inside, and three sets of ion selective electrode sheets 22, 23, 24 which are attached to the side surfaces and bottom surface of the block 21. On the upper surface of the block 21, an opening 25a for receiving a spotted sample solution and an opening 25b for receiving a spotted reference solution are provided, and a bridge member (fiber bridge) 26 for electrically connecting the received sample solution with the received reference solution is fixed. On each of the side surfaces and bottom surface of the block 21, a pair of openings 27a, 27b for supplying separately the received sample solution and the received reference solution onto the ion selective electrode sheet.

The block 21 is made of water-impermeable, electri-

cally insulating material such as polystyrene. The fiber bridge 26 is made from polyethylene terephthalate thrown fibers.

Each of the ion selective electrode sheets 22, 23, 24 is composed of an non-electroconductive support, a silver layer, a silver chloride layer, an electrolytic material layer and an ion selective membrane, as shown in Fig. 1. The ion selective electrode sheets 22, 23, 24 preferably have ion selective membranes different from each other, so that ionic activities of plural ionic components are measured in one device. The ion selective electrode sheet 22 is fixed to the surface of the block 21 under such condition that the uppermost ion selective membrane 221 faces the openings 27a, 27b and the electrode terminals 222 are extended from the edges of the block 21.

In Fig. 4, (1) indicates a upper surface of the hollow block 21, (2) is a front view, and (3) is a side view. In Fig. 4, the inner hollow space of the block 21 has a pair of solution reservoirs 30a, 30b which are separated by a partition 31 to supply the sample solution received in the opening 25a and the reference solution received in the opening 25b separately to the solution-supplying openings 27a, 28a, 29a, and the solution-supplying openings 27b, 28b, 29b, respectively.

The block 21 typically has a dimension of 4 mm x 20 mm x 4 mm (height), as illustrated in Fig. 4. Each of the solution-receiving openings 25a, 25b has a dimension of 3 mm x 3 mm. Each of the solution-supplying openings 27a, 27b has a dimension of 2 mm x 2 mm, while each of the solution reservoirs 30a, 30b has a dimension of 3 mm x 3 mm x 3.5 mm.

The structure of the block 21 is not limited to that illustrated in Fig. 4. For instance, the partition 31 is a plate arranged in the inner hollow space at the center position. Otherwise, a part of the partition 31 extends

to both side portions into the reservoirs 30a, 30b so as to reduce the volume of the reservoir, as is shown in Fig. 5. Alternatively, a pair of porous materials 32a, 32b are placed independently in both reservoirs to transmit the sample solution and the reference solution, separately, as is shown in Fig. 6.

Each of the solution-supplying openings 27a, 27b, 28a, 28b can have a slanting ceiling from the inside to the outside, as is shown in Fig. 7, so that air bubbles are not produced in the opening.

The upper surface of the block 21 can have a groove for protecting the fiber bridge 26 and further for reducing a required volume of the solutions. The fiber bridge 26 can be placed and fixed in the groove.

The ion activity-measuring device of the invention is not limited to the above-mentioned structure having three sets of the ion selective electrode sheets. For instance, only two sets of the ion selective electrode sheets 22, 23 are attached to the side surfaces, as is illustrated in Fig. 8-(1). Otherwise, four sets of ion selective electrode sheets 22, 23, 24, 33a (in combination with 34b) are attached to the side surfaces and the bottom surface, as is illustrated in Fig. 8-(2).

Alternatively, seven sets of ion selective electrode sheets 42a, 42b, 42c, 43a, 43b, 43c, 44 attached to the block 41 on the upper surface, bottom surface, the side surfaces, as is illustrated in Fig. 9. On each of the upper surface and the bottom surface, the electrode sheets 42a to 42c, and 43a to 43c are placed in parallel. On the upper surface of the block 42 are arranged the solution-receiving openings 45a, 45b and the fiber bridge 46. The electrode sheets attached to the upper surface and bottom surface can be two sets or four sets in place of the above-mentioned three sets. The electrode sheets may be not attached to the side surfaces, or may be attached only one of the upper surface and bottom surface.

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The ionic activity-measuring device can be composed of a vertically extended block 51 and several sets of ion selective electrode sheets 52a, 52b, 52c, ---, 53a, 53b, 53c, ----, 54, which are attached to the side surfaces and the bottom surface, as is illustrated in Fig. 10. On the side surfaces, the electrode sheets are arranged in parallel. On the upper surface of the block 51 are provided the solution-receiving openings 55a, 55b and the fiber bridge 56. The ion selective electrode sheets attached to the side surface can be two or three sheets. The electrode sheet may not be attached to the bottom surface, and may be attached to one side surface only.

The ionic activity-measuring device can be composed of a block 61 in a drum form and three sets of ion selective electrode sheets 62, 63, 64 which are attached to the side surfaces and the bottom surface, as is illustrated in Fig. 11. On the upper surface of the block 61, two solution-receiving openings 65a, 65b and one fiber bridge 66 are placed. The ion selective electrode sheets attached to the side surface and the bottom surface can be two, four or more sheets.

The ionic activity-measuring device can be composed of a block 71 composed of two drums united to each other and two sets of ion selective electrode sheets 72a, 72b, 73a, 73b, as illustrated in Fig. 12. In Fig. 12, the electrode sheet is composed of a pair of single electrode 72a (73a) and single electrode 72b (73b), which are separated from each other. On the upper surface of the block 71 are provided solution-receiving openings 75a, 75b and a fiber bridge 76. Three or more electrode sheets can be attached to the side surface, and an ion selective electrode sheet can be attached to the bottom surface.

The drum structure can be replaced with other structures such as polygonal pillar such as pentagonal pillar, hexagonal pillar, and octagonal pillar. On the side surfaces and/or bottom surfaces can be attached ion se-

lective electrode sheets.

The preparation of the ionic activity-measuring device of the invention is described by referring to the device of Fig. 2.

5 First, plastic material such as polyolefin (e.g., polystyrene) is molded to produce a block of Fig. 3.

Independently, three sets of ion selective electrode sheets each of which has a different ion selective membrane are prepared in one of known manners describe, for
10 instance, in United States Patents No. 4,053,381, No. 4,171,246, No. 4,219,968, and Research Disclosure, No. 16113 (1977, September). A representative ion selective electrode sheet is illustrated in Fig. 3.

The three sets of ion selective electrode sheets are
15 attached to the side surfaces and the bottom surface of the block using a double adhesive-coated tape. On the upper surface, a fiber bridge made of a polyethylene terephthalate thrown fiber is placed to connect the solution-receiving openings and fixed to the surface using
20 appropriate adhesive.

The ionic activity-measuring devices can be manufactured in a mass scale, and a plurality of devices can be conveyed and stored after the devices 20 are encased in a frame 34 made of plastic material, as is shown in Fig.
25 13-(1). For the use of the encased device, each device is separated together with its casing, and placed in a potentiometer for the desired measurement.

For manufacturing the vertically extended ionic activity-measuring device of Fig. 10, two block units
30 511, 512 (551a, 551b, 552a, 552b: solution-receiving opening, 571a, 571b: solution-supplying opening) are produced in several numbers and jointed together in the vertical direction to give the desired vertically extended block 51.